QP CODE: 105326	Reg. No:
Q. 00DL. 1000E0	1109: 110:

First Year Pharm D Degree Examinations July 2019

Pharmaceutical Inorganic Chemistry

Time: 3 Hours Total Marks: 70

- Answer all Questions.
- Draw Diagrams wherever necessary.

Essays (3x10=30)

- 1. Explain the principle, reaction and procedure involved in the limit test of iron and sulphates.
- 2. Explain the different types of precipitation titrations
- 3. What are neutralization titration and explain the theories of acid base concept

Short notes: (8x5=40)

- 4. Explain electrolyte combination therapy
- 5. Explain the role of flourides as anti anticaries agents
- 6. What are acidifiers. Explain the test for purity, identification test and medicinal uses of sodium phosphate and ammonium chloride
- 7. What is standard solution, primary standard and secondary standard and explain the requirements for the substances used as primary standard.
- 8. What are masking and demasking agents in complexometric titration and give some examples.
- 9. Explain why boric acid cannot be directly titrated against alkali. Mention the medicinal uses of boric acid.
- 10. Explain the different types of errors
- 11. Explain the physiological role of iron
