QP CODE:212006 (NEW SCHEME) Reg.I	No:
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Second Year B.Pharm Degree Supplementary Examinations December 2017 PHARMACEUTICAL ANALYSIS

(2012 Scheme)

Time: 3 Hours Total Marks: 100

- Answer all Questions.
- · Write equations wherever necessary.

Essay (3x10=30)

- 1. What is cerimetry. Explain the standardization methods of secondary standards used in redox titrations
- 2. Explain various steps involved in gravimetric analysis with suitable examples
- 3. What is an error and mention its types and methods to control them

Short notes (14x5=70)

- 4. Explain Karl Fischer method
- 5. Relation between importance of buffer and stability of complexes
- 6. Calcium assay by gravimetry
- 7. Explain acid-base concepts and law of mass action
- 8. Explain the theoretical basis, types of solvents and limitations of non-aqueous titrations
- 9. Theory of indicators in neutralization titrations
- 10. How do drugs containing primary aromatic amines assayed
- 11. Explain the assay of oxygen as per IP
- 12. Enumerate different techniques employed in analysis of drugs
- 13. Preparation and standardization method for 0.1M EDTA
- 14. Explain adsorption indicator method in precipitation titrations
- 15. How does mixture of metals assayed
- 16. Differentiate pH and pM indicators
- 17. Common ion effect and ionic product of water
