2012 Scheme

QP CODE: 111006 Reg. No:

First Year B. Pharm Degree Supplementary Examinations January 2024 Pharmaceutical Chemistry - I

(Inorganic & Physical Chemistry)

Time: 3 Hours Total Marks: 100

- Answer all questions to the point neatly and legibly Do not leave any blank pages between answers • Indicate the question number correctly for the answer in the margin space
- Answer all parts of a single question together Leave sufficient space between answers
- Write equations wherever necessary.

Essay (3x10=30)

- 1. Explain the mechanism of maintenance of physiological acid base balance. Elaborate the method of preparation and medicinal uses of sodium chloride
- 2. Give the radiopharmaceutical preparations, clinical application, dosage, hazards and storage of lodine -131
- 3. Give the preparation, properties, assay, storage and uses of ferrous sulphate.

Short notes (14x5=70)

- 4. Enumerate the composition and medicinal uses of oral rehydration salts.
- 5. What are astringents. Give the preparation, assay, storage and uses of Zinc chloride.
- 6. Explain the acid neutralizing capacity of an antacid. Give the preparation and uses of aluminium hydroxide gel.
- 7. Give the identification test, storage and uses of sodium citrate.
- 8. Comment on the role of flourides as anti-caries agents. Give the preparation and uses of sodium fluoride.
- Note on dimethicone.
- 10. Give the preparation, properties, storage and uses of nitrous oxide.
- 11. Explain the physiological role of iron and chloride
- 12. Give the preparation, properties, assay and uses of Boric acid
- 13. Explain the theory of coordination compounds and mention the uses of EDTA & 1, 10-phenanthroline.
- 14. Limit test for chlorides.
- 15. Discuss sodium methyl paraben and sodium lauryl sulphate and mention their uses
- 16. Give the identification test, test for purity and uses of iron and ammonium citrate
- 17. What are sedatives. Give the identification test, storage and uses of potassium bromide
