2012 Scheme

QP CODE: 112006

First Year B. Pharm Degree Supplementary Examinations June 2023

Pharmaceutical Chemistry - II

(Organic Chemistry)

Time: 3 Hours

- Answer all questions to the point neatly and legibly Do not leave any blank pages between answers • Indicate the question number correctly for the answer in the margin space
- Answer all parts of a single question together

 Leave sufficient space between answers
- Write equations wherever necessary.

Essav

- 1. With the supporting evidence, elaborate the Kekule structure of Benzene.
- 2. What is Kharach effect. Explain antimarkovnikov's rule. Write briefly on stability of conjugated dienes.
- 3. Explain the polarity of carbonyl group. Give two important methods of preparation of carbonyl compounds. Give the mechanism involved in the aldol condensation and Cannizaro's reaction.

Short notes

- 4. List any three general methods of preparation of alcohols.
- 5. Give the mechanism involved in Kolbe-Schmidt and Fries rearrangement.
- 6. Explain the ring stabilities of cyclohexane with suitable examples.
- 7. Briefly explain the acidity of nitro compounds with suitable examples.
- 8. What is decarboxylation reaction. Give any two important methods of preparation of carboxylic acids.
- 9. Give brief account on malonic and acetoacetic esters. Write their synthetic applications.
- 10. Give three reactions for aromatic nitro compounds.
- 11. Explain the concept of resonance with suitable illustrations.
- 12. Explain the role of solvents in nucleophilic aliphatic substitution reactions.
- 13. Write the structures for the following IUPAC names.
 - a) 4-hydroxy Octane
 - b) 3-chloro-5-methyl-7-nitro Decane
 - c) 3-oxo-Heptanoic acid
 - d) Hex-4-en-2-one
 - e) Ethyl ethanoate
- 14. What is Walden inversion. Explain with suitable example.
- 15. Explain Hoffmann degradation of amides with suitable example.
- 16. Give an account of acidity of phenols. Explain the effect of electron withdrawing groups on the acidity of phenols.
- 17. Explain Saytzeff's rule with suitable examples.

Total Marks: 100

Reg. No:

(14x5=70)

(3x10=30)