QP Code: 322006 Reg. No......

Third Semester B. Pharm Degree Regular/Supplementary Examinations October 2022

Physical Pharmaceutics I

(2017 Scheme)

Time: 3 Hours Max. Marks: 75

- Answer all questions to the point neatly and legibly Do not leave any blank pages between answers
 Indicate the question number correctly for the answer in the margin space
- Answer all parts of a single question together Leave sufficient space between answers.
- Draw diagrams wherever necessary

Essays (2x10=20)

- 1. Explain the different methods for the determination of average particle size and particle size distribution. Explain in detail about the sieving technique.
- 2. Explain the various factors affecting the solubility of gases in liquids.

Short Notes (7x5=35)

- 3. Summarize the solubility of partially miscible liquids.
- 4. Glassy state.
- 5. Explain in short about eutectic mixture.
- 6. Explain micromeretics. Write its application in pharmaceutical sciences.
- 7. Elaborate about hydrogen bonded complexes.
- 8. Demonstrate any two methods for the determination of true density.
- 9. Buffer capacity.

Answer Briefly (10x2=20)

- 10. Define the term Buffer.
- 11. What are the various factors influencing the pH of buffer solution.
- 12. Draw a neat label diagram of one component system.
- 13. Critical point.
- 14. Enlist the scientific ways of expressing particle size distribution.
- 15. Particle number.
- 16. Classify the metal ion coordination complexes.
- 17. What are clathrates.
- 18. Define Raoult's law.
- 19. The terms used for expressing solubility.
