QP Code: 122006 Reg. No......

First Semester B. Pharm Degree Regular/Supplementary Examinations April 2024

Pharmaceutical Analysis - I

(2017 Scheme)

Time: 3 Hours Max. Marks: 75

- Answer all questions to the point neatly and legibly Do not leave any blank pages between answers
 Indicate the question number correctly for the answer in the margin space
- Answer all parts of a single question together Leave sufficient space between answers
- Draw table/diagrams/flow charts wherever necessary

Essay (2x10=20)

- 1. Explain the principle involved in Polarography. Write about Ilkovic equation, construction and working of dropping mercury electrode
- 2. What are the concept and type of oxidation and reduction titration. Give in detail on three of them

Short Notes (7x5=35)

- 3. Write in detail about the preparation & standardization of sulphuric acid.
 - 4. Explain conductometric titrations
 - 5. Co-precipitation and post precipitation
 - 6. Write on calomel electrode and its applications
 - 7. Titration of calcium gluconate
 - 8. Write about estimation of Sodium benzoate by non aqueous titration
 - 9. Explain neutralization titration curves

Answer Briefly (10x2=20)

- 10. With example give properties of Aprotic solvents
- 11. What are masking and demasking agents.
- 12. Sources of errors.
- 13. Properties of primary and secondary standards.
- 14. What is mean by reference and indicator electrodes. Give examples.
- 15. Differentiate molarity and molality
- 16. Name the theories of acid-base indicators
- 17. End point detection of precipitation titrations.
- 18. Differentiate volhards and modified volhards method
- 19. Explain the principle of diazotisation titration with one example
