First B.Sc Optometry Degree Supplementary Examinations - March 2013

CHEMISTRY

Answer all guestions Draw diagram wherever necessary

Essay:

Time: 3 hrs

- 1. Describe the term chemotherapy and chemotherapeutic agents. What do you understand by term sulpha drugs and mention the method of synthesis and its uses.
- 2. Discuss in detail the optical isomerism of tartaric acid and differentiate between optical and geometrical isomerism.

Short notes

- 3. Discuss E1 and E2 mechanism of elimination reactions.
- 4. Discuss briefly liquid solid adsorption chromatography.
- 5. What are hormones. Explain their biological importance taking suitable examples.
- 6. What are carbohydrates and how they are classified.
- 7. Predict the product of the reactions:
 - CH₂=CHCI +HCI.
 - CH3_C=C_(CH3)2 +H20/H+.
 - Benzene+Secondary butyl chloride
 - ,. CH3CH2CH2CI+Alc.KOH .
 - propylene+HBr.

Answer briefly

- (10x2=20)8. Discuss the methods that can be used to differentiate between geometrical isomers taking the example of maleic and fumaric acid.
- 9. Give the structures of ascorbic acid and cyanocobalamin
- 10. The mechanism of Friedel Crafts reaction.
- 11. Discuss the stereochemistry of SN1 reactions.
- 12. What are emulsions and how they are classified.
- 13. Mention any two reactions of glucose.
- 14. Draw the resonance structure of nitrobenzene.
- 15. How will you get cellulose from its sources.
- 16. Outline the important uses of benzene.
- 17. Tyndall effect.

Fill in the blanks

- 18. Rayon is regenerated form of
- 19. Hybridization of carbon in benzene is
- 20. The nitrating mixture helps the generation of the electrophile______ in the nitration of benzene.
- 21...The conversion of enantiomer into the racemic mixture is called
- 22. . is a colloid in which the dispersion medium is a liquid and dispersed phase is a solid.

Q.P.Code 104013

(5x5=25)

(5x1=5)

(2x15=30)

Max marks : 80

Reg. No.:....