Biochemistry Time: 3 Hrs Max. Marks:80 Answer all questions Draw diagram wherever necessary • $(2 \times 10 = 20)$ 1. Explain the important Buffer systems in regulating the body pH $(2\frac{1}{2} + 2\frac{1}{2} + 2\frac{1}{2} + 2\frac{1}{2} = 10)$ 2. Enumerate the glass wares used in clinical biochemistry. Mention the use and care of any 4 items. How do you clean the used and new glass ware. (2+4+4=10)Short notes:

- 3. Explain the steps for the preparation of exactly 0.1 Ν NaOH solution by secondary titration
- 4. Principle and use of colourimeter
- 5. Compare the different models of atomic structure.
- 6. Explain the significance in laboratory practice about water of crystallization, hygroscopic and deliquescent salts.
- 7. Principle and use of sensitivity and specificity in clinical laboratory practice.
- 8. Blood gas analysis

Answer briefly:

- 9. Ionization of water, H ion concentration and pH.
- 10.Use of centrifuge
- 11. Nutritional importance of lipids.
- 12. Liver function tests
- 13. Principle of serum electrophoresis. Name one abnormal band seen.
- 14. Biochemical changes seen and causes of metabolic acidosis
- 15. How do you evaluate the limits of error in a laboratory.
- 16. Normal values of common six electrolytes in blood.
- 17. Calorific value of carbohydrates, proteins and lipids.
- 18. Enumerate the renal function tests, the functions tested and significance.

First Year B.Sc Perfusion Technology Degree Examinations, August 2014

Q.P. CODE: 103016

Essays:

(6x5=30)

(10x3=30)

Reg. No: