

## ERRATA/MODIFICATIONS IN REGULATIONS AND SYLLABUS

### M.Pharm Pharmaceutical Chemistry (MPC) Course Code – 277

Sl. No.	Location	Error	Correction/Modification
<b>ERRATA/MODIFICATIONS IN REGULATIONS</b>			
1	Page 4 <b>Table -3: Course of study for M.Pharm III &amp; IV Semester Semester IV</b>	Discussion /Final Presentation	Presubmission Discussion / Presentation
2	Page 6 <b>Table - 5:</b> (Guidelines for awarding credit points for co curricular activities; Note)	<b>International conference: Held outside India</b>	<b>International conference - Conference in which resource persons from two or more nations or nationalities participate.</b>
3	Page 6 <b>Table - 5:</b> (Guidelines for awarding credit points for co curricular activities; Note)	<b>International journal - The Editorial board outside India</b>	<b>International journal - one quarter of editorial board are from an outside nation/ one third of papers published originate from an outside nation</b>
4	Page 6 Below Table - 5 <i>* (Asterisk)</i>	*The credit points assigned for extracurricular & co-curricular activities shall be given by the Principals of Colleges and the same shall be submitted to the University. The criteria to acquire this credit points shall be defined by the colleges from time to time.	*The credit points assigned for co-curricular activities shall be given by the Principals of Colleges and the same shall be submitted to the University along with the attendance and marks scored by the candidates in semester IV.
5	Page 34 <b>2.16 Project work to be done if any;</b> (first paragraph)	4 copies of the project work shall be submitted (typed & bound copy not less than 75 pages)	4 copies of the project work shall be submitted (typed & bound copy not less than 75 pages and not more than 150 pages) <i>Format for printing the thesis is given separately.</i>
6	Page 34 <b>2.16 Project work to be done if any;</b> (second paragraph)	The internal and external examiner appointed by the University shall evaluate the project at the time of the Practical examinations of other semester(s).	The internal and external examiner appointed by the University shall evaluate the project.

7	Page 35 <b>2.16. Project work to be done if any; Evaluation of Dissertation work</b> (Mark distribution)	<b>Evaluation of Dissertation work</b> Objective(s) of the work done - 50 marks Methodology adopted – 150 marks Results and Discussion - 250 marks Conclusion and outcomes - 50 marks Total - 500 marks	<b>Evaluation of Dissertation work</b> Objective(s) of the work done - 25 marks Methodology adopted - 75 marks Results and Discussion - 100 marks Conclusion and outcomes - 50 marks Total - 250 marks
8	Page 35 <b>2.16. Project work to be done if any; Evaluation of Presentation</b> (Mark distribution)	<b>Evaluation of Presentation</b> Presentation of work - 100 marks Communication skills - 50 marks Question and answer skills - 100 marks Total - 250 marks	<b>Evaluation of Presentation</b> Presentation of work - 75 marks Communication skills - 25 marks Question and answer skills - 50 marks Total - 150 marks
9	Page 35 <b>2.21 Logbook</b>	Registers to be maintained	Registers to be maintained: The Institutions have to maintain registers for student activities such as Seminar/Assignment (Semester I & Semester II), Journal club presentataion (Semester III & Semester IV), Minutes of Mentoring committee (Semester III & Semester IV), and Co-curricular activities (Semester I - IV) in addition to the mandatory general records as specified by the University from time to time.
10	Page 36 <b>3.3. Scheme for examination showing maximum marks and minimum marks</b>	The end semester examination for each theory and practical Course through semester I to IV shall be conducted by the respective University except fot the subject with asterisk symbol (*)	The word “ <b>respective</b> ” stands omitted.
11	Page 37 <b>Table-7: Schemes for internal assessments and end semester examination; SEMESTER I;</b> (Second Last Row)	Seminar/Assignment: Total marks 100	Total marks 100; split up as 25 marks for each theory subject out of which 10 marks for seminar and 15 marks for assignment
12	Page 37 <b>Table-7: Schemes for internal assessments and end semester examination; SEMESTER II;</b> (Second Last Row)	Seminar/Assignment: Total marks 100	Total marks 100; split up as 25 marks for each theory subject out of which 10 marks for seminar and 15 marks for assignment

13	Page 38 <b>Table-8: Schemes for internal assessments and end semester examinations SEMESTER III</b> (Third row)	Discussion/Presentation (proposal presentation) <b>50</b> <b>50</b>	Discussion/Presentation (Proposal presentation) <b>25</b> <b>25</b>
14	Page 38 <b>Table-8: Schemes for internal assessments and end semester examinations SEMESTER III</b> (Last row)	<b>Total - 525</b>	<b>Total - 500</b>
15	Page 38 <b>Table-8: Schemes for internal assessments and end semester examinations SEMESTER IV</b>	Discussion/Presentation (Proposal presentation)	Presubmission Discussion / Presentation
16	Page 39 <b>Allowed to keep terms (ATKT):</b> (Second paragraph)	However, he/she shall not be eligible to attend the courses of IV semester until all the courses of I, II and III semesters are successfully completed.	However, he/she shall not be eligible to submit the thesis until all the courses of I, II and III semesters are successfully completed.
<b>ERRATA IN SYLLABUS</b>			
1	Page 8 2.6. Syllabus <b>MPC 101 T</b> objectives - 3	Theoretical and practical skills of the instruments	Theoretical and practical skills for handling of the instruments
2	Page 9 2.6. Syllabus <b>MPC 101T</b> (4 Chromatography)	i) i) Gel Chromatography	i) Gel Chromatography
3	Page 11 2.6. Syllabus <b>MPC 102T</b> (1 Basic Aspects of Organic Chemistry)	Organic intermediates:	Reactive intermediates:

4	Page 11 2.6. Syllabus <b>MPC 102T</b> (1 Basic Aspects of Organic Chemistry)	Addition reactions a) Nucleophilic uni-and bimolecular reactions (SN1 and SN2) b) Elimination reactions (E1 & E2; Hoffman & Saytzeff's rule) c) Rearrangement reaction	a) Addition reactions b) Nucleophilic uni-and bimolecular substitution reactions (SN1 and SN2) c) Elimination reactions (E1 & E2; Hoffman & Saytzeff's rule) d) Rearrangement reactions
5	Page 21 <b>MPC 201T</b> (1 UV and IR spectroscopy)	Wood ward – Fieser	Woodward-Fieser